# **COMMENT ON THE ARTICLE «SOLID CaCO3 FORMATION IN WATERS OF** CIRCULATING COOLING SYSTEMS OF POWER PLANTS UNDER THE **CONDITIONS OF ELECTRIC LOAD CHANGE»** [Water and water purification technologies. Scientific and technical news. Vol. 23, № 2 (2018), pp. 12-21. DOI: https://doi.org/10.20535/2218-93002322018144957]

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In article [Kochmarskii, 2018] in the analysis of the Ist stability index and the rate of the formation of calcium carbonate  $D_{ve}$  from circulated water (CW) of the cooling systems of electric power stations with cooling towers under changeable electric load Ne the direct dependence of dynamic parameter  $\boldsymbol{\varphi}$  on Ne has not been taken into account. In some cases this led to inadequate reflection of  $I_{st}(Ne)$  and  $D_{ve}(Ne)$  dependences. This inadequacy has been taken into account and proper laws of the expected value have been obtained. It was shown that the rate of the formation of calcium carbonate  $D_{ve}$  is proportional to the electric load of the electric power station, to the concentration of  $Ca^{2+}$  ions in feed water and to the fraction of the heat used which is dispersed by evaporation at cooling towers, and it is inversely proportional to the water volume of circulated cooling system (CCS) and the average efficiency (AE) of the station.

Key words: Calcium carbonate; circulated water; condensation power stations; electric load; evaporation; rate of CaCO<sub>3</sub> formation; stability index.

#### Introduction

In a previous article [Kochmarskii, 2018] the actual task of calculating stability index  $I_{st}(Ne)$  and formation rate of CaCO<sub>3</sub> from waters of CCS of electric power stations (EPS), in particular such ones, which are equipped with cooling towers, under conditions of changeable electric load were discussed. However, in [Kochmarskii, 2018] in modeling this regime the dependence of dynamic parameter  $\varphi$ on electric load which led to the inadequate reflection of Ist(Ne) and Dve(Ne) dependences have not been taken into account.

In this article for modeling dependences  $I_{st}(Ne)$  and  $D_{ve}(Ne)$  we use basic formulas (3), (6), (7), (8a), (9), (11) and (12), obtained in [Kochmarskii, 2018]. For a stable regime of CCS and the established law of CaCO<sub>3</sub> formation from CW, see (11) from [Kochmarskii, 2018], taking into account the evident dependence  $\phi(Ne)$  which as it follows from (1) and is related to the expenditure of CW and evaporation G<sub>ev</sub>(Ne),

$$\varphi(\text{Ne}) = \frac{G_{\text{f}}}{G_{\text{f}} - G_{\text{ev}}(\text{Ne})}; \quad I_{\text{st}}(\text{Ne}) = \frac{1}{1 + \varphi(\text{Ne}) \cdot T_{\text{f}} \cdot \text{K} \cdot C_{\text{HCO3}\infty}}; \quad T_{\text{f}} = \frac{V}{G_{\text{f}}}. \quad (1)$$

$$D_{ve}(Ne) = \frac{dC_{Cass}}{dt} = \frac{C_{Ca0}}{T_{f}} \cdot \left[1 - I_{st}(Ne)\right] = \frac{\phi(Ne) \cdot K \cdot C_{Ca0} \cdot C_{HCO300}}{1 + \phi(Ne) \cdot K \cdot T_{f} \cdot C_{HCO300}},$$
(2)

G<sub>f</sub>, G<sub>ev</sub> - correspondingly, the expenditure of water for feeding and evaporation, m<sup>3</sup>/hour; C<sub>Cass</sub> molar concentration of solid CaCO3 that is formed from CW, g-mole/m3; CHCO300, CCa0concentrations of bicarbonate and calcium ions in stationary CW and in feed water, (g-ions)/m<sup>3</sup>; K - kinetic coefficient of the reaction of forming CaCO<sub>3</sub> under the interaction of Ca<sup>2+</sup> and HCO<sub>3</sub><sup>-</sup> ions

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#### Mathematical modeling and optimization

in the adopted in [Kochmarskii, 2018] model of the formation of calcium carbonate. K is determined by a special procedure for water of acting or designed CCS. For non-stabilized by inhibitors waters K =  $(1.5...4.4)\cdot10^{-4}$ m<sup>3</sup>/(g-ions)·hour, for stabilized ones it may be three-four times less [Gayevskii, 2018]. It is shown in [Kochmarskii, 2018] that the direct dependence of water expenditure for evaporation on electric load is given by the expressions

$$G_{ev}(Ne) = K_{ev} \cdot A(\eta) \cdot Ne, \quad A(\eta) = 3600 \cdot \frac{(1/\eta - 1)}{\rho_{w} \cdot r_{w}}, \quad \eta = \frac{Ne}{\sum \frac{Ne_{i}}{\eta_{i}}}$$
(3)

 $K_{ev}$  - coefficient of evaporation is equal to the heat part formed in a turbine condenser which is transferred in cooling towers by evaporation, this value depends on the temperature of CW and surrounding conditions is changed [Laptiev, Vedgaieva, 2004] within boundaries of  $0.30 < K_{ev} < 0.87$ ;  $\rho_w$ ,  $r_w$  - correspondingly, water density and specific heat of evaporation under temperature of CW; Ne =  $\Sigma N_i$  - total electric load of the station, MW;  $\eta$  - AE at the station; A( $\eta$ ) - quantity of evaporated water,  $m^3$  per 1MW hour of heat dispersed by cooling towers, A = (2.3...3.1)  $m^3/(MW \cdot hour)$ .

Coefficient of evaporation  $K_{ev}$  from (3) may be calculated on the basis of the mathematical model of the cooling process in a cooling tower, or determined by a special procedure [Laptiev, Vedgaieva, 2004]. Other values in (3) are tabular, or belong to such ones which are measured by the control regulation of the regime of stations operation.

#### Simulation

Using the expression (3) and  $\varphi(Ne)$  from (1), after transforming (1) and (2) we get expected dependences

$$I_{st}(Ne) = \frac{G_{f} - K_{ev} \cdot A(\eta) \cdot Ne}{G_{f} - K_{ev} \cdot A(\eta) \cdot Ne + V \cdot K \cdot C_{HCO3oo}}; (4)$$
$$D_{ve}(Ne) = G_{f} \cdot \frac{K \cdot C_{Ca0} \cdot C_{HCO300}}{G_{f} - K_{ev} \cdot A(\eta) \cdot Ne + V \cdot K \cdot C_{HCO3oo}}. (5)$$

As in [Kochmarskii, 2018] we use obtained here expressions (4) and (5) to analyze the operation of EPS [Kochmarskii at al, 2014] in CCS of which partial softening of feed water, cooling towers and also inhibitors of deposits are used. For calculations we adopt: Ne = (300...1200)MW;  $\phi_{00} = 2.8...3.8$ ; V =  $(200...280) \cdot 10^3$ m<sup>3</sup>; K =  $(1.1...4.4) \cdot 10^{-4}$ m<sup>3</sup>/(g-ions) · hour; C<sub>Ca0</sub> = 0.7(g-ions)/m<sup>3</sup>, C<sub>HCO300</sub>=7(g-ions)/m<sup>3</sup>, A( $\eta$ ) = 2.63m<sup>3</sup>/MW · hour. Calculation results are shown below.

We see from fig. 1, with the increased parameter K (decreased dosage of CaCO<sub>3</sub> inhibitor), which controls the rate of CaCO<sub>3</sub> formation, see curves 1...4, stability index of CW is reduced expectedly. Such behavior of stability index is the same with the increased electric load Ne. That is, the operation of EPS under the maximum load is accompanied by the maximum decreased stability of circulated water. However, from fig. 2 we see that the stability of CW is increased with the growth of AE of the station. Hence, the decreased stability of CW due to the growth of load of EPS is partially compensated by its AE increase, see fig. 2.

Interesting is the dependence of  $I_{st}(G_f)$  shown in fig. 3. The stability of CW increases with the increased consumption of feed water.

This conclusion is not evident and it is to be considered while using the recirculation of the blowthrough which decreases the consumption of fresh water in CCS. That is, in using such an approach it is necessary to consider the possible decreased stability of CW. The dependence  $I_{st}(Kev)$  is shown in fig. 4. We see that the operation of CCS under increased coefficient of evaporation leads to decreased stability index. Such regime is implemented under high temperatures of CW (in summer). It is evident that the winter period of CCS operation contributes to the high stability index.



Fig. 1. Dependence of stability index on electric load. Curves 1...4 correspond to value of constant K = (1.1; 2.2; 3.3; 4.4) $\cdot$ 10<sup>-4</sup> m<sup>3</sup>/(gions)·hour. Value of other parameters:

 $\eta=0.35;~\rho_w=998 kg/m^3,~r_w=2.47 MJ/kg;~V=2.4\cdot 10^5 m^3;~G_f{=}~3000 m^3/hour.$ 



Fig. 3. Dependence of stability index on expenditure of replenished water. Ne = 800MW;  $\eta = 0.35$ . Values of other parameters are the same as in fig.1.



Fig. 2. Dependence of stability index on AE at Ne = 800MW. Values of other parameters are the same as in fig.1.



Fig. 4. Dependence of stability index on evaporation coefficient. Ne=800MW;  $\eta = 0.35$ ;  $G_f = 2000m^3$ /hour. Values of other parameters are the same as in fig.1.

For example, by the data of fig. 4 we find that during the transition from winter operation regime to summer one, the stability index of CW decreases from 1.6 to 3.0 times. This fact is to be taken into account in planning the regime of stabilizing CW by inhibitors or in developing other measures of optimization of CCS operation regimes.

Let us analyze the behavior of the formation rate of CaCO<sub>3</sub> from CW under different operation conditions of CCS. Calculations will be done by formula (5). Results are shown in fig. 5 and 6. From fig. 5 we see that the growth of the capacity of a station from 300 to 1200MW with K=1.1.10<sup>-4</sup>m<sup>3</sup>/(g-ions)·hour is accompanied by the increased rate of CaCO<sub>3</sub> formation by 5 times, see curve 1 in fig.5, with K =  $2.2.10^{-4}$ m<sup>3</sup>/(g-ions)·hour - by 2.75 times, with K =  $3.3.10^{-4}$ m<sup>3</sup>/(g-ions)·hour - by 2.4 times and with K =  $4.4.10^{-4}$ m<sup>3</sup>/(g-ions)·hour by 2.2 times. That is, with great K the formation rate of CaCO<sub>3</sub> formation by 5 times. From fig. 6 we see that with the increased load of stations. The dependence D<sub>ve</sub>( $\eta$ ) is also noticeable. From fig. 6 we see that with the increased AE within the range of 0.3 ... 0.4 the rate of CaCO<sub>3</sub> formation decreases by 1.6 times. Comparing these dependences in fig. 5 and 6 we see that to compensate for the growth of the formation rate of calcium carbonate under the increased loading is impossible by the growth of CE.

Dependences in fig.6 correlate with curves in fig. 3 and expressions (1) and (2) because the stability growth of CW is always accompanied by the decreased rate of CaCO<sub>3</sub> formation.

The decreased rate of CaCO<sub>3</sub> formation with the AE growth, see fig. 6, is explained by the fact that under the same capacity the growth of AE is accompanied by the expenditure of smaller quantity of water for evaporation, and *under the stable feeding of CCS* its blow-through grows ( $G_f - G_{ev} = G_e$ ) and, correspondingly, the concentration of Ca<sup>2+</sup> and HCO<sub>3</sub><sup>-</sup> ions capable to generate the deposition of CaCO<sub>3</sub> decreases, see fig. 7.



Fig. 5. Dependence of the rate of CaCO<sub>3</sub> formation on electric power loading Ne. Curves 1...4 correspond to the value of constant K such as is in fig.1. Values of other parameters:  $\eta = 0.35$ ;  $\rho_w = 998 \text{kg/m}^3$ ;  $r_w = 2.5 \text{MJ/kg}$ ;  $V=2.4 \cdot 10^5 \text{m}^3$ ;  $G_f = 3000 \text{m}^3$ /hour.





Fig. 6. Dependence of the rate of  $CaCO_3$  Fig. 7. Dependence of parameter  $\varphi$  on AE of formation on AE. Ne = 800MBT. Values of block, see (1) and (3). Values of other other parameters are the same as in fig. 5.

We see that the growth of AE within the range of 30 ... 40% is accompanied by the decreased concentrations of ions by  $\approx 1.6$  times.

Hence, in analyzing conditions of CCS operation and in calculating the rate of the formation of deposits it is necessary to consider changes in operation regimes of CCS due to the change of stations load which is accompanied by the change of evaporation of CW in cooling towers. This means that adequate mathematical model of CCS operation must consider the value of electric load and water regime of stations operation.

### Summary

- 1. Basic parameters reflecting dependences  $I_{st}(Ne)$  and  $D_{ev}(Ne)$  are water consumption for CCS feeding, AE of station  $\eta$  and its electric load Ne. Besides,  $I_{st}(Ne)$  and  $D_{ev}(Ne)$  parameters depend on the quality of CW and coefficient K that characterizes the rate of CaCO<sub>3</sub> formation.
- 2. With increased electric load of the station the stability index of CW decreases and the rate of CaCO<sub>3</sub> formation grows. On the contrary, with the growth of AE the stability of CW grows and the rate of CaCO<sub>3</sub> formation and the level of salts concentration reduce.
- 3. With decreased coefficient of evaporation  $K_{ev}$  (CCS operation under decreased temperatures), stability index of CW grows, correspondingly, the rate of CaCO<sub>3</sub> formation reduce.

- 4. The proposed here methods of calculating I<sub>st</sub>(Ne) and D<sub>ev</sub>(Ne) allow us to make a quantitative forecast of CW stability in designing power stations on the basis of data about the quality of water at the local source of feeding and to substantiate the selection of optimum measures concerning the minimization of the rate of CaCO<sub>3</sub> formation.
- 5. For the reliable forecast of CW, stability for new stations it is necessary to have experimental studies concerning the determination of coefficient K in water that models CW of designed CCS.
- 6. Expressions (1) (5) may be the basis for the technical and economic optimization of the process of minimizing deposits under the changeable CCS operation regimes, as well as for developing adequate mathematical models of automating the process of deposits control under such conditions.

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# ЗАУВАЖЕННЯ ДО СТАТТІ «ВИДІЛЕННЯ ТВЕРДОГО СаСО<sub>3</sub> З ВОД ОБОРОТНИХ СИСТЕМ ОХОЛОДЖЕННЯ ЕЛЕКТРОСТАНЦІЙ В УМОВАХ ЗМІННОГО ЕЛЕКТРИЧНОГО НАВАНТАЖЕННЯ» [Water and water purification technologies. Scientific and technical news. T. 23, № 2 (2018), pp. 12-21. DOI: https://doi.org/10.20535/2218-93002322018144957]

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В [Kochmarskii, 2018] при аналізі індексу стабільності  $I_{st}$  та швидкості виділення карбонату кальцію  $D_{ve}$  з оборотної води (OB) систем охолодження електростанцій з градирнями при змінному електричному навантаженні Ne не було враховано явної залежності динамічного параметра  $\varphi$  від Ne. Це призвело до неадекватного відображення залежностей  $I_{st}(Ne)$  та  $D_{ve}(Ne)$ .

В даній роботі ця неточність врахована і отримані коректні закони поведінки індексу стабільності I<sub>st</sub> та швидкості виділення карбонату кальцію D<sub>ve</sub>. Показано, що швидкість виділення карбонату кальція D<sub>ve</sub> пропорційна електричному навантаженню електростанції; концентрації іонів Ca<sup>2+</sup> у воді підживлення та частці відпрацьованого тепла, що розсіюється випаровуванням на градирнях; обернено пропорційна водному об'єму оборотних систем охолодження та середньому ККД станції. Індекс стабільності OB з ростом електричного навантаження зменшується, проте, зі збільшенням ККД станції стабільність OB зростає. Чисельне моделювання для конденсаційної електростанції з електричним навантаженням (300...1200) MBm, спорядженої градирнями, показало, що ріст стабільності OB завдяки росту ККД не компенсує його зменшення внаслідок збільшення навантаження станції. При роботі станції в умовах підвищеного випаровування (літній сезон) стабільність OB знижується. Підвищення подачі води підживлення при сталих режимах випаровування та виділення СаСОз сприяє збільшенню стабільності OB (завдяки зменшенню коефіцієнту концентрування солей).

Отримані тут вирази  $I_{st}(Ne)$  та  $D_{ve}(Ne)$  дозволяють розрахувати необхідні дози інгібіторів відкладень та розробити процедури для досягнення заданої стабільності OB. Вони будуть корисними при проектуванні нових станцій для покращення кількісного прогнозу стабільності OB на підставі даних про якість води місцевого джерела живлення. Їх можна застосовувати при обґрунтуванні вибору оптимальних заходів щодо мінімізації швидкості виділення CaCO<sub>3</sub> в умовах змінного навантаження. Вирази  $I_{st}(Ne)$  та  $D_{ve}(Ne)$  будуть корисними при розробці матмоделей автоматизації процесу контролю над відкладеннями.

**Ключові слова:** випаровування; карбонат кальцію; індекс стабільності; оборотна вода; швидкість виділення; електричне навантаження.

# ЗАМЕЧАНИЕ К СТАТЬЕ «ВЫДЕЛЕНИИ ТВЕРДОГО СаСО<sub>3</sub> ИЗ ВОД ОБОРОТНЫХ СИСТЕМ ОХЛАЖДЕНИЯ ЭЛЕКТРОСТАНЦИЙ В УСЛОВИЯХ ПЕРЕМЕННОЙ ЭЛЕКТРИЧЕСКОЙ НАГРУЗКИ» [Water and water purification technologies. Scientific and technical news. T.23, № 2 (2018), pp. 12-21. DOI: https://doi.org/10.20535/2218-93002322018144957]

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В [Kochmarskii, 2018] при анализе индекса стабильности  $I_{st}$  и скорости выделения карбоната кальция  $D_{ve}$  с оборотной воды (OB) систем охлаждения электростанций с градирнями при переменной электрической нагрузке Ne явно не была учтена зависимость параметра  $\varphi$  от Ne. Это привело в некоторых случаях к неадекватному отображению зависимостей  $I_{st}(Ne)$  и  $D_{ve}(Ne)$ .

Здесь эта зависимость учтена и получены корректные количественные законы поведения искомых величин. Показано, что скорость выделения карбоната кальция  $D_{ve}(Ne)$  пропорциональна электрической нагрузке электростанции, концентрации ионов  $Ca^{2+}$  в воде подпитки и части отработанной теплоты, рассеиваемой испарением в градирнях, обратно пропорциональна водному объему оборотной системы охлаждения и среднему КПД станции. Ключевые слова: испарение; карбонат кальция; индекс стабильности; оборотная вода; скорость выделения СаСО3; электрическая нагрузка.